

Determination of ^{93}Mo (and ^{94}Nb) in nuclear decommissioning waste from a nuclear reactor

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RAS academic meeting (23 April 2018)

Nutech meeting (26 April 2018)

18th Radiochemical Conference (13-18 May 2018, Mariánské Lázně)

NKS RadWorkshop (8-12 October 2018, Risø)



DTU Nutech

Center for Nukleare Teknologier

A short introduction

- DTU Nutech (Technical University of Denmark, Center for Nuclear Technologies) is the Danish competence center for nuclear technologies.
- The 3 former Danish research reactors were on the campus.
- DTU Nutech (1956-2006 called Risø) has long-term experience on radiochemical analyses of (among others) nuclear waste, especially decommissioning waste.
- ^3H , ^{14}C , ^{36}Cl , ^{41}Ca , ^{55}Fe , ^{59}Ni , ^{63}Ni , ^{90}Sr , ^{93}Mo , ^{93}Zr , ^{94}Nb , ^{99}Tc , ^{129}I , ^{210}Po , ^{210}Pb , ^{226}Ra , ^{237}Np , ^{234}U , ^{235}U , ^{236}U , ^{238}U , ^{238}Pu , ^{239}Pu , ^{240}Pu , ^{241}Pu , ^{241}Am , ^{244}Cm ; ^{60}Co , ^{152}Eu , ^{154}Eu , ^{134}Cs , ^{137}Cs .

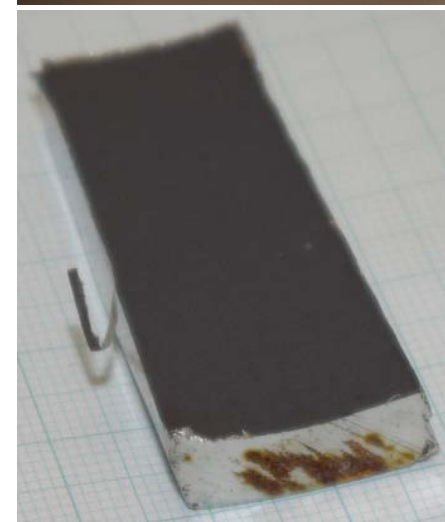
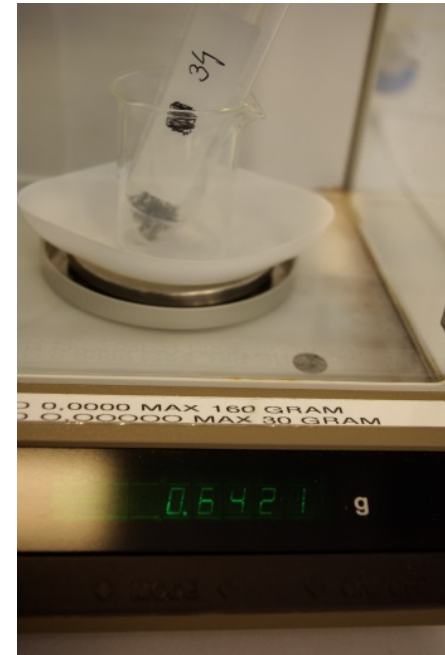


Samples

- Metals from a NPP (under decommissioning)
 - Main radioactive components: ^{60}Co (\approx kBq-MBq) + ^{55}Fe
 - Induced activity samples
 - Small pieces, irradiated by neutrons
 - Activation products
 - Surface layer activity samples
 - Big pieces, contacted with primary water
 - Corrosion products

- Model sample (for method development):
 - NIST Standard Reference Material 123c
 - (Cr-Ni-Nb Stainless Steel; AISI 348)

Metal	m/m %
Fe	the rest (68.52%)
Cr	17.40%
Ni	11.34%
Mn	1.75%
Nb	0.65%
Mo	0.22%
Co	0.12%



12.10.2018

Goal

- To develop a new method for determination of ^{93}Mo and ^{94}Nb in nuclear power plant decommissioning wastes

^{93}Mo

- $t_{1/2} = (4.0 \pm 0.8) \times 10^3$ years
- Electron capture
- Possibilities for detection:
 - X-ray spectrometry: 16.5 keV (62%) and 18.6 keV (9%) – K_{α} and K_{β} lines of Nb
 - LSC: Auger-electrons
 - MS: presence of $^{\text{nat}}\text{Mo}$ (abundance sensitivity of ^{92}Mo (15%) and ^{94}Mo (9%))

^{94}Nb

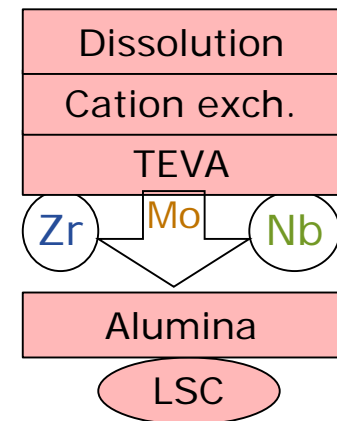
- $t_{1/2} = 2.0 \times 10^4$ years
- β^{-} - γ emitter ($E_{\beta, \text{max}} = 470$ keV)
- Detection by gamma-spectrometry:
 - 703 keV (98%) and 871 keV (100%)
- Radiochemical separation is needed before measurement!
 - $^{93}\text{Mo}/^{60}\text{Co} \approx 10^{-5} - 10^{-3}$
 - $^{93}\text{Mo}/^{93\text{m}}\text{Nb} \approx 10^{-5} - 10^2$
 - $^{94}\text{Nb}/^{60}\text{Co} \approx 10^{-5} - 10^{-3}$

(activity ratios in our samples)



Overview of our method

- Dissolution
- Combined chromatographic separation
 - Cation exchange
 - TEVA
 - Alumina
- Measurements



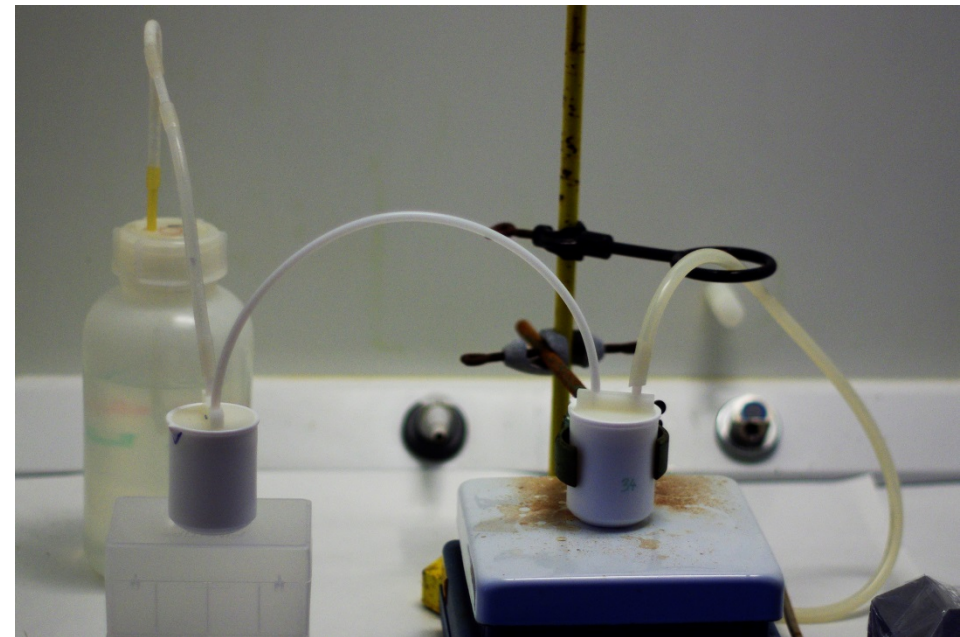
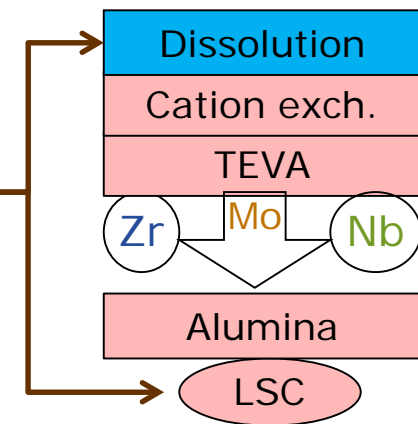
Detection techniques used in method development:

- Gamma-spectrometry: ^{60}Co , ^{94}Nb , ^{125}Sb ; $^{99\text{m}}\text{Tc}$
- ICP-OES: stable elements (Fe, Cr, Ni, Mn, Mo, Nb, Zr)
 - Interferences
 - Extra problem: elimination of HF by dilution, evaporation or complexation (H_3BO_3)



Dissolution

- Surface samples: “leaching” of the activity from the surface
- Induced samples: direct dissolution
- Addition of carriers (stable Mo and Nb)
 - ICP-OEC measurement of aliquots taken before and after separation → Recovery
- Dissolution and repeated evaporation using aqua regia (68% HNO₃ + 36% HCl) and 40% HF
 - Oxidation to MoO₄²⁻ (+VI, crucial)
 - Green solution: Cr³⁺, Ni²⁺
- Dissolution in 0.1 M HF
- Dilution until 0.02 M HF



1. column: Cation exchange resin. Getting rid of the matrix

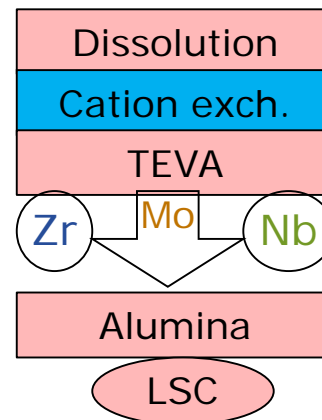
Load & rinse:
0.02 M HF
(lower $c \rightarrow$ higher DF)

Retained: cations
(majority of the activity)

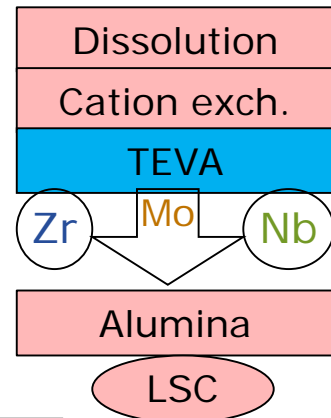
$^{54}\text{Mn}^{n+}$, $^{55}\text{Fe}^{3+}$, $^{60}\text{Co}^{2+}$,
 $^{59}\text{Ni}^{2+}$, $^{63}\text{Ni}^{2+}$, $^{65}\text{Zn}^{2+}$, Cr^{3+}

Pass through: anions

^{93}Zr : ZrF_6^{2-}
 ^{125}Sb : SbF_6^-
 ^{99}Tc : TcO_4^-
 $^{93\text{m}}\text{Nb}$ and ^{94}Nb : NbF_6^- , NbOF_5^{2-}
 ^{93}Mo : MoO_2F_3^- , $[\text{MoO}_2\text{F}_4]^{2-}$, MoF_7^- , MoOF_5^-
 CrO_4^{2-} (When applying reducing agents,
the Mo recovery is reduced as well.)

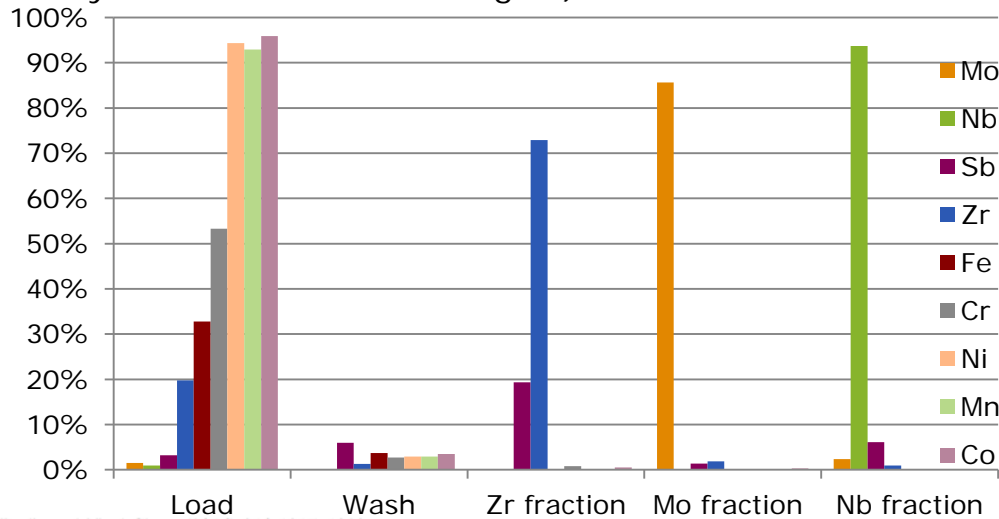


2. column: TEVA® resin. Separation of anions



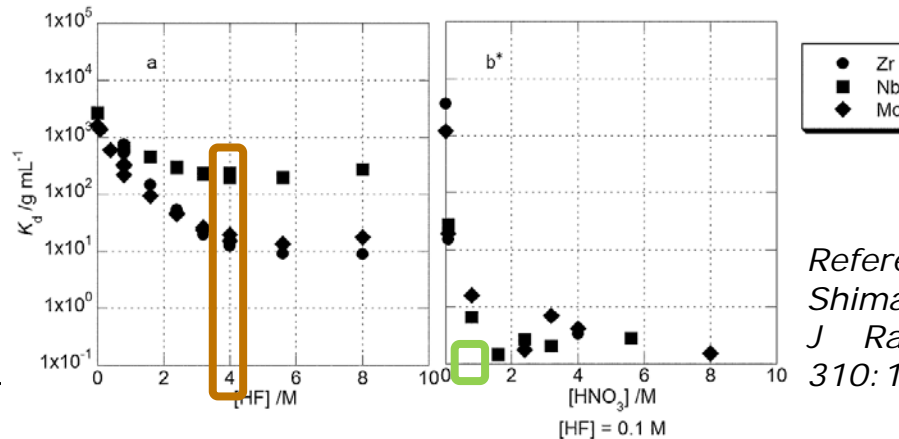
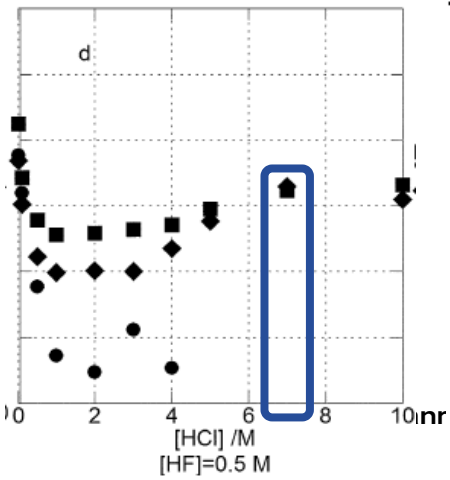
- Based on a quaternary ammonium salt (TrisKem product, very similar to anion exchangers)

- Load & rinse: 0.02 M HF
- Zr strip:** + **Sb**
5 mL 7 M HCl/0.5 M HF
- Mo strip:** 12 mL 4 M HF + **Sb** contamination
- Nb strip:** 10 mL 1 M HNO₃



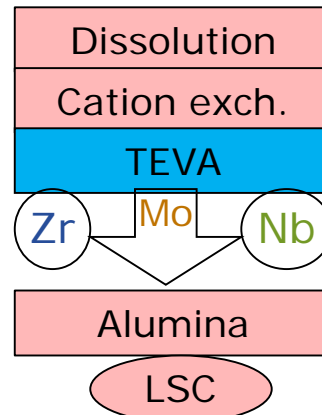
J Radioanal Nucl Chem (2016) 310:1317–1323

40 mm column 3 mL fractions			
	Zr	Mo	Nb
1.	104%	59%	91%
2.	2%	7%	0.2%
3.	0%	2%	0.1%
4.	0%	1%	-
5.	0%	1%	-
	+ 1% Nb	+ 1% Nb	



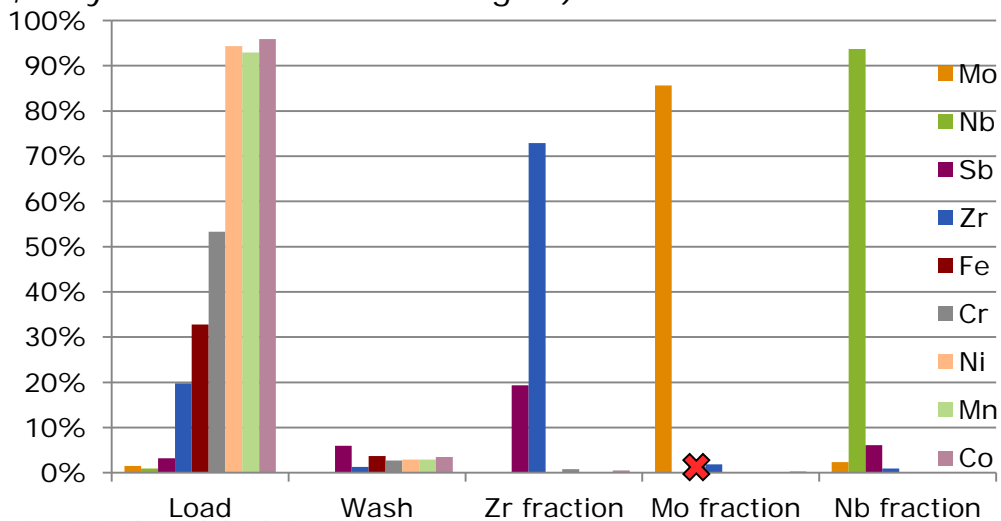
Reference:
Shimada & Kameo (2016)
J Radioanal Nucl Chem
310: 1317-1323

2. column: TEVA® resin. Separation of anions



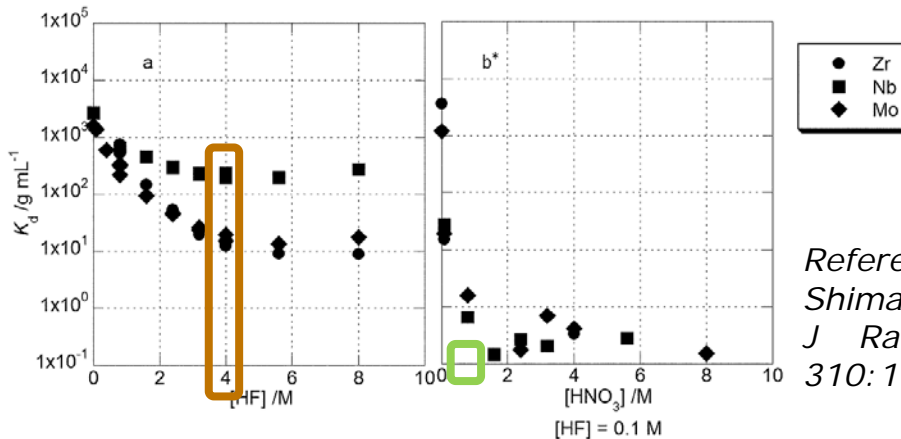
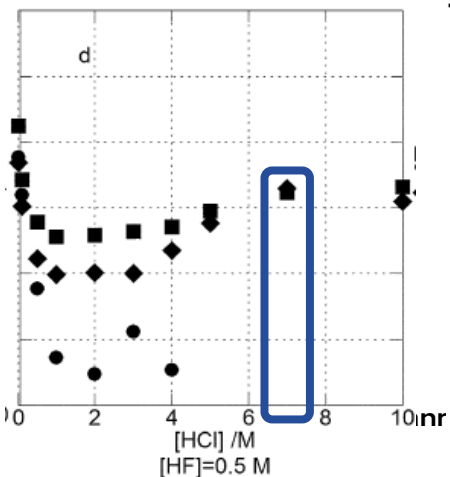
• Based on a quaternary ammonium salt
(TrisKem product, very similar to anion exchangers)

- Load & rinse: 0.02 M HF
- **Zr strip:** + **Sb**
5 mL 7 M HCl/0.5 M HF
- 3 mL 0.02 M HF
- **Mo strip:** 12 mL 4 M HF
no **Sb** contamination
- **Nb strip:** 10 mL 1 M HNO₃

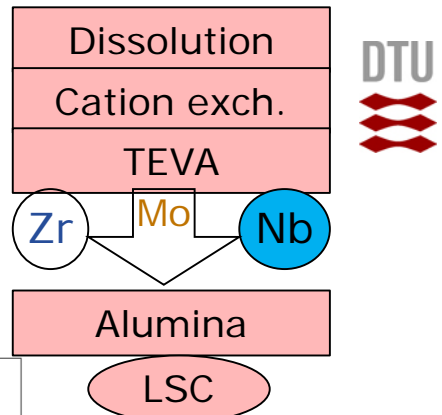


J Radioanal Nucl Chem (2016) 310:1317–1323

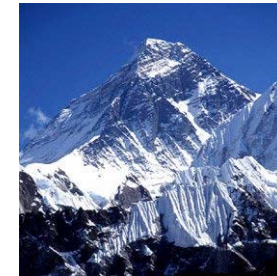
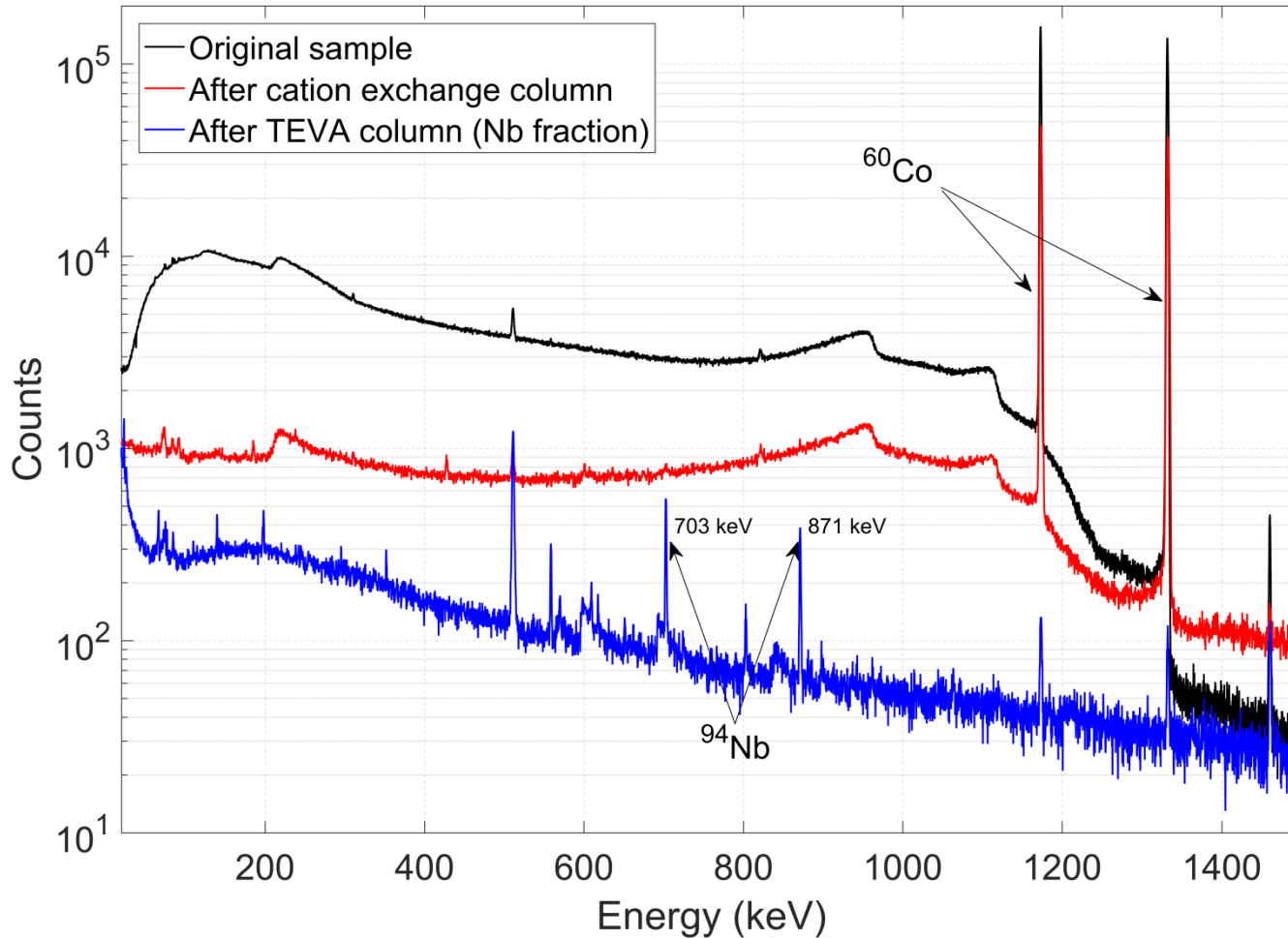
40 mm column 3 mL fractions			
	Zr	Mo	Nb
1.	104%	59%	91%
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3.	0%	2%	0.1%
4.	0%	1%	-
5.	0%	1%	-
	+ 1% Nb	+ 1% Nb	



Reference:
Shimada & Kameo (2016)
J Radioanal Nucl Chem
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Measurement of ^{94}Nb using HPGe



10^5 Bq
 ^{60}Co
< 70 Bq
 ^{94}Nb

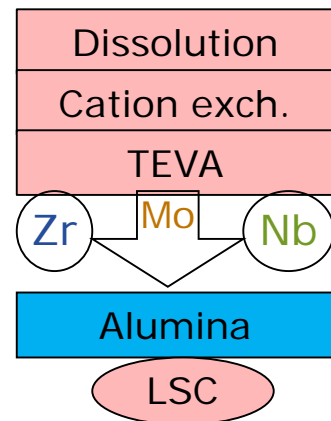


2×10^2 Bq
 ^{60}Co
< 0.2 Bq
 ^{94}Nb



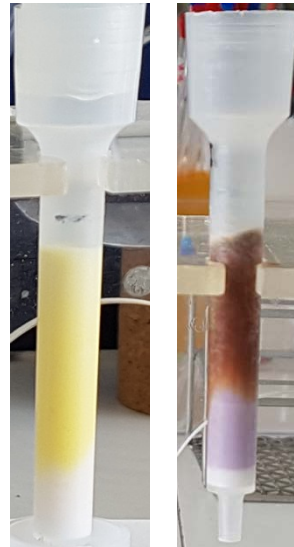
10^{-1} Bq
 ^{60}Co
0.2 Bq
 ^{94}Nb
(MDA = 0.04 Bq)

3. column: Alumina (Al_2O_3). Purification of Mo



- Widely applied for $^{99\text{m}}\text{Tc}/^{99}\text{Mo}$ separation in “technogenerator”s (using HNO_3 media)
- But practically no information is available about usage of HF media

- Load & rinse: 1 M HNO_3
- Wash: 0.1 M HNO_3
 H_2O
0.01 M NH_3
- Mo strip: ≥ 1 M NH_3



- Load & rinse: ≤ 0.1 M HF
- Wash: H_2O
0.01 M NH_3
- Mo strip: ≥ 1 M NH_3

Other metals pass mainly through

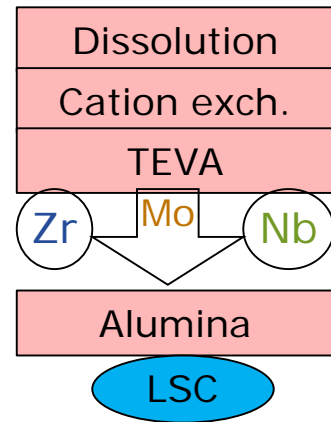
Other metals are retained

- In general, results with HNO_3 and HF are similar
- Higher c of $\text{NH}_3 \rightarrow$ more effective elution of Mo
- Sb always contaminates the Mo – except when cc. NH_3 is used for elution

Reference:

Bernhard (1994)

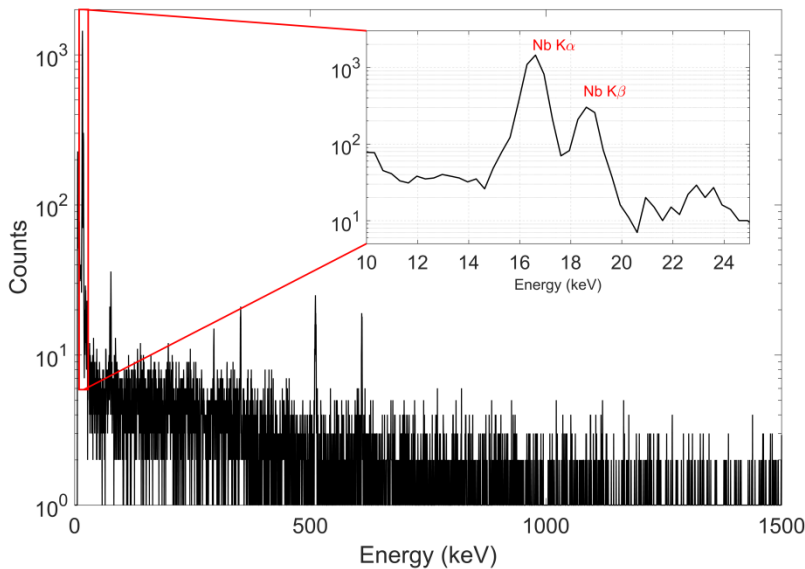
J Radioanal Nucl Chem 177(2): 321-325



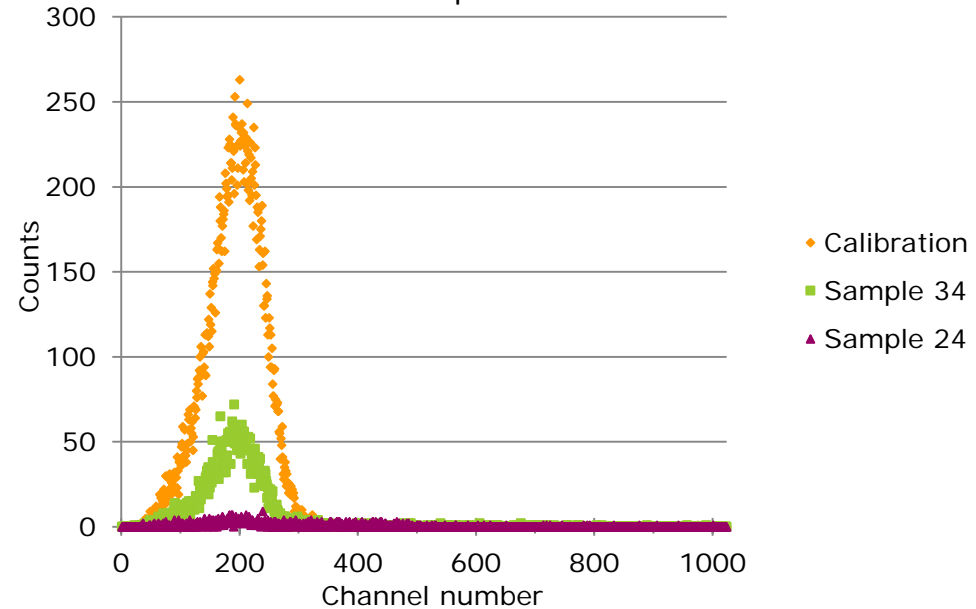
Measurement of ^{93}Mo using LSC

- Evaporated sample ($\approx 450 \mu\text{L dw}$) + 20 mL Ultima Gold LLT
- How to calibrate? Remember Per's presentation
- $\eta = 52\%$

X-ray spectrum (HPGe)
Problem: self-absorption







LSC spectrum
Problem: quench



Method performance

- Recoveries (model samples):
 - Mo: typically over 85%
 - Nb: typically over 75%
 - Zr: typically over 70%
- Analysis of real samples is in progress
- Decontamination factors:

Separation of Mo			
Element	Cation exchange	TEVA Mo fr.	Alumina
Fe	$\geq 10^3$	10^3	$\geq 4 \cdot 10^2$
Cr	≈ 2	500	$\geq 8 \cdot 10^3$
 Co	10^3	10^4	$\geq 10^2$
Ni	10^3	10^4	
Mn	10^3	10^4	
 Nb	1	$5 \cdot 10^2$	$\geq 2 \cdot 10^4$
Zr	1	≥ 10	$\geq 7 \cdot 10^2$
 Sb	1	$\geq 10^3$	$\geq 10^2$
Tc	1	$3 \cdot 10^2$	$4 \cdot 10^2$

Separation of Nb		
Element	Cation exchange	TEVA Nb fr.
Fe	$\geq 10^3$	10^5
Cr	≈ 2	10^3
 Co	10^3	10^4
Ni	10^3	10^4
Mn	10^3	10^4
Mo	1	10^2
Zr	1	$\geq 10^2$
Sb	1	
Tc	1	10^3

Summary. Conclusions and perspectives

- A method for determination of ^{93}Mo (and ^{94}Nb) - based on combined chromatographic separation - was developed
- Model samples: recoveries and separation factors are satisfying
- Real samples: in some cases very low recoveries - maybe Zr saturates the TEVA column? (longer columns should be used)
 - Comparison of results with estimated values (based on modelling)
- Validation by “standard addition” method
- Gamma-spectrometric measurement of ^{94}Nb before chemical separation (in the presence of lots of ^{60}Co) using an anti-coincidence gamma-spectrometer
- Method might be extended for determination of Zr (ICP: recovery, LSC: activity)



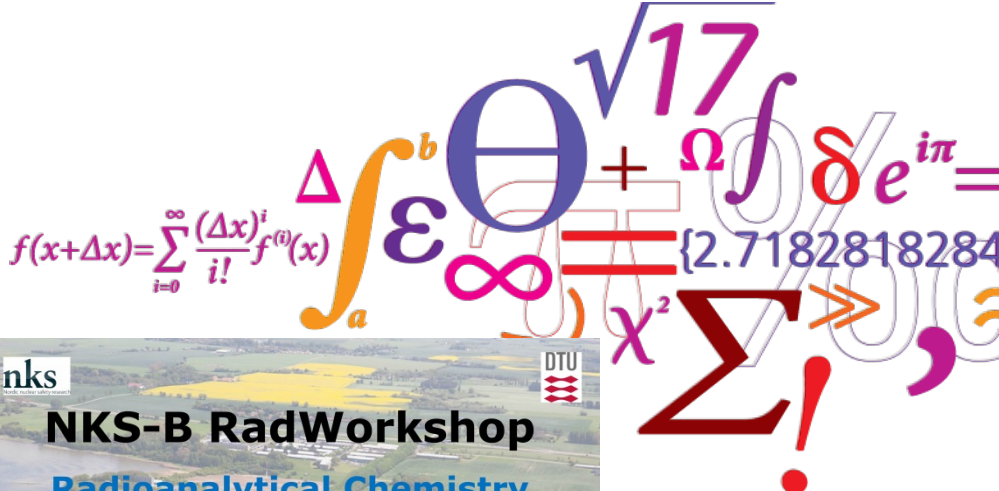
Thank you very much

for all your help and kind attention.

<http://www.nutech.dtu.dk/english>



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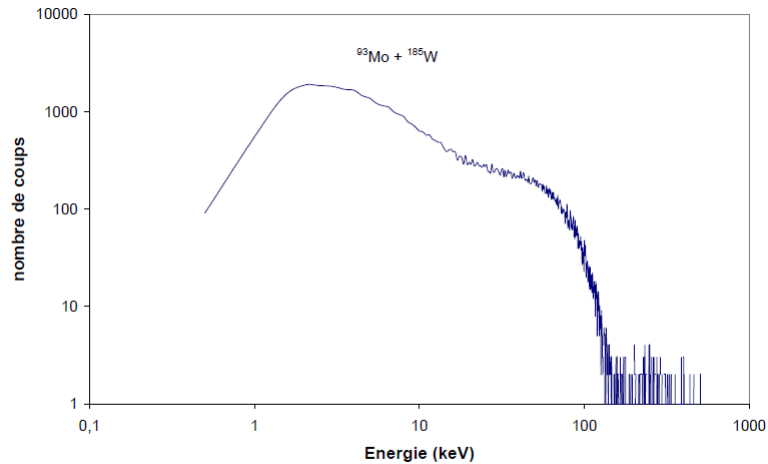
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NKS-B RadWorkshop

Radioanalytical Chemistry for Nuclear Decommissioning and Waste Management

8-12 OCTOBER 2018
DTU Risø Campus, Roskilde, Denmark

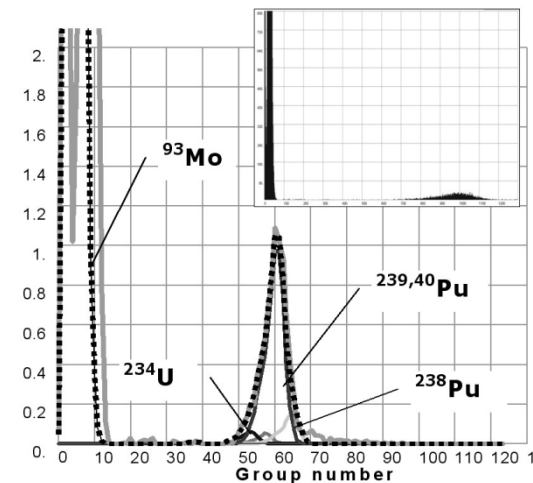
We acquired the 3rd LSC spectrum in the world about ^{93}Mo



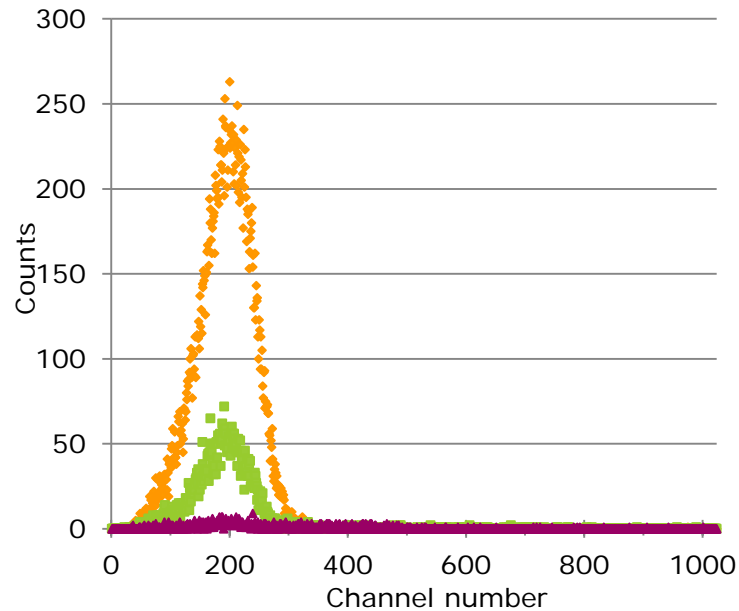
Reference:
 Bombard (2005)
 PhD Thesis,
 Nantes University,
 France (pp 139)

Contaminated with
 ^{185}W (433 keV,
 100% β^-)

Figure 62 : Spectres de scintillation liquide de la fraction 202ME_M (temps de comptage = 600 minutes).



Reference:
 Ermakov et al.
 (2005)
 In: Chalupnik,
 Schönhofer, Noakes
 (eds):
 LSC 2005, Advances
 in Liquid Scintillation
 Spectrometry (pp
 89–98)



◆ Calibration
 ■ Sample 34
 ▲ Sample 24

Calibration of LSC for measurement of ^{93}Mo

- No certified ^{93}Mo can be purchased
- “Home-made” solution: Separation of Mo from irradiated Nb
 - Dissolution and repeated evaporation: 40% HF + 68% HNO_3
 - Dissolution in 6 M HF
 - First separation step: Precipitation of Nb_2O_5 and co-precipitation on $\text{Fe}(\text{OH})_3$ (using NH_3)
 - Based on “the lost method” from Patricia Puech (1998): Détermination des radionucléides zirconium 93 et molybdène 93 dans des effluents de retraitement des combustibles irradiés. Thesis, Univ. Paris XI, 211, France*
 - Repeated evaporation: 36% HCl + 68% HNO_3
 - Repeated evaporation: 40% HF
 - Dissolution in 0.1 M HF
 - Dilution until 0.05 M HF
 - Second separation step: purification on Alumina column
- Performance of separation: Recovery of Mo $\approx 60\%$
DF of Nb $\geq 10^6$
- Measurement by calibrated X-ray spectrometer
- Measurement by LSC