

Historical overview and current challenges in radionuclide analyses in different waste matrices and materials at SCK•CEN, Belgium

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Introduction

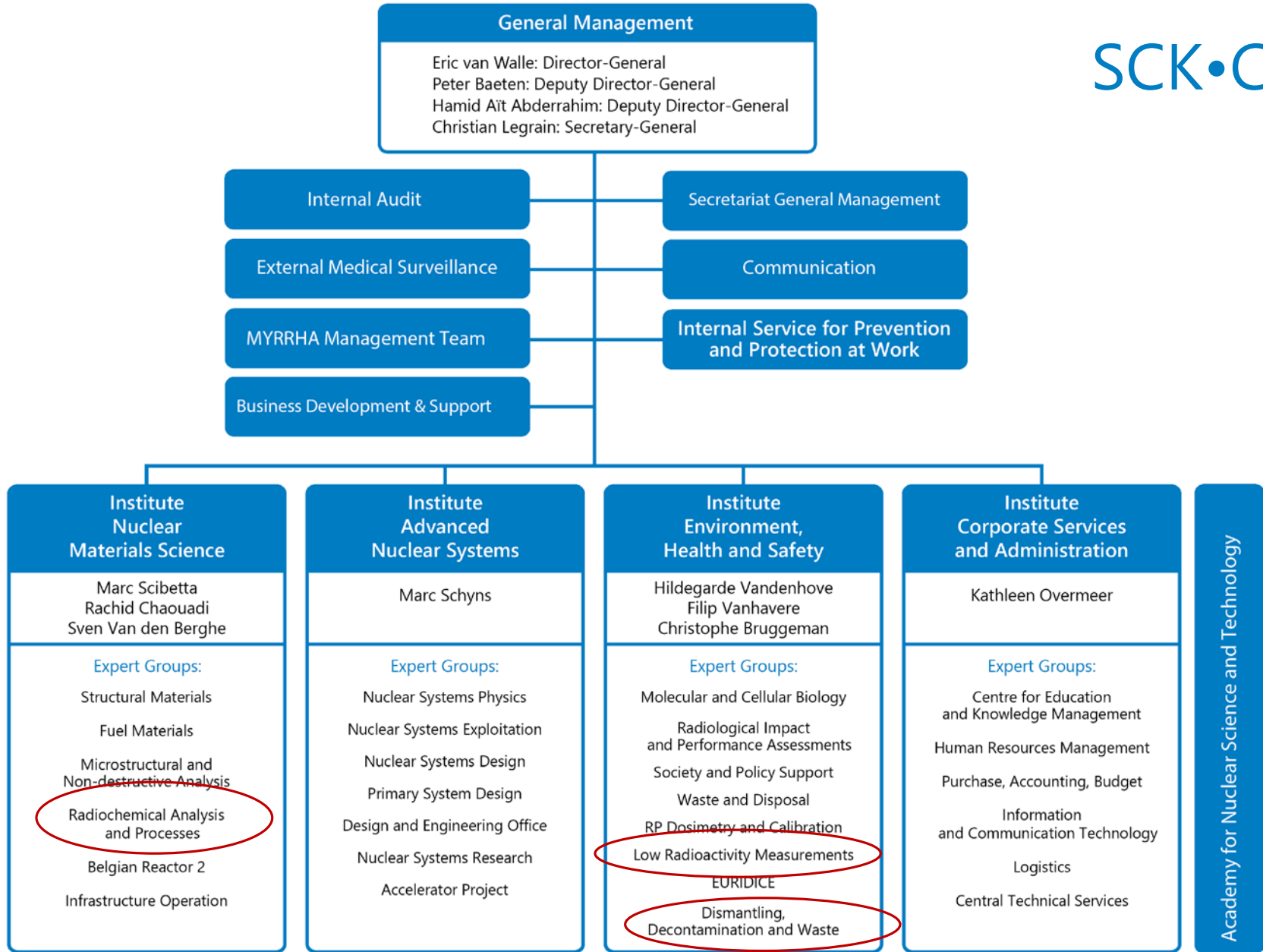
What has been done in the past

Current challenges

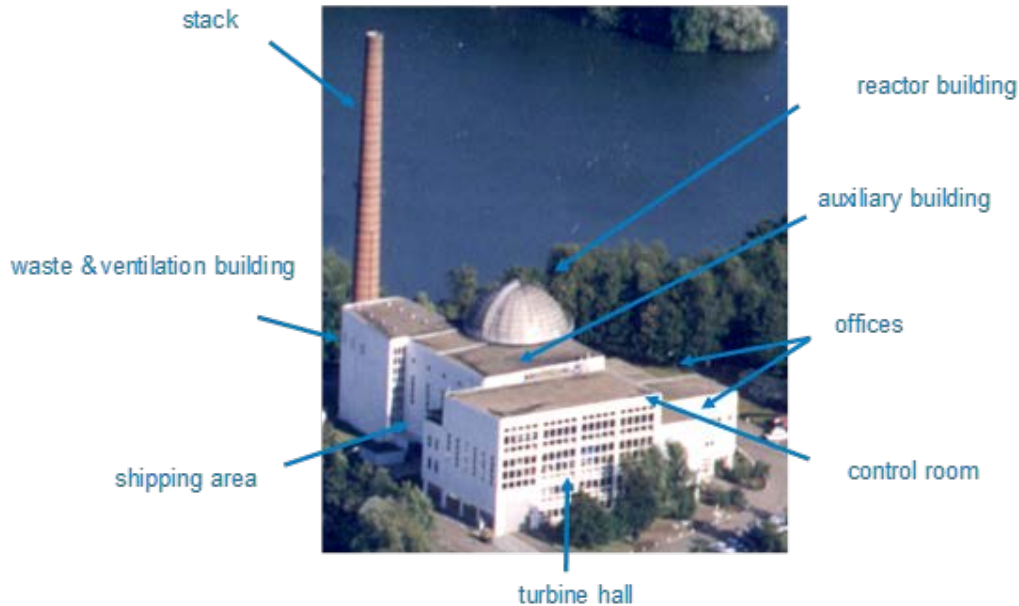
- ✓ Determination of ^{14}C , ^{36}Cl and ^{129}I (RCA)
- ✓ Determination of ^{99}Tc in environmental samples (LRM)

Future plans

Conclusions



BR3 (Belgian Reactor 3)

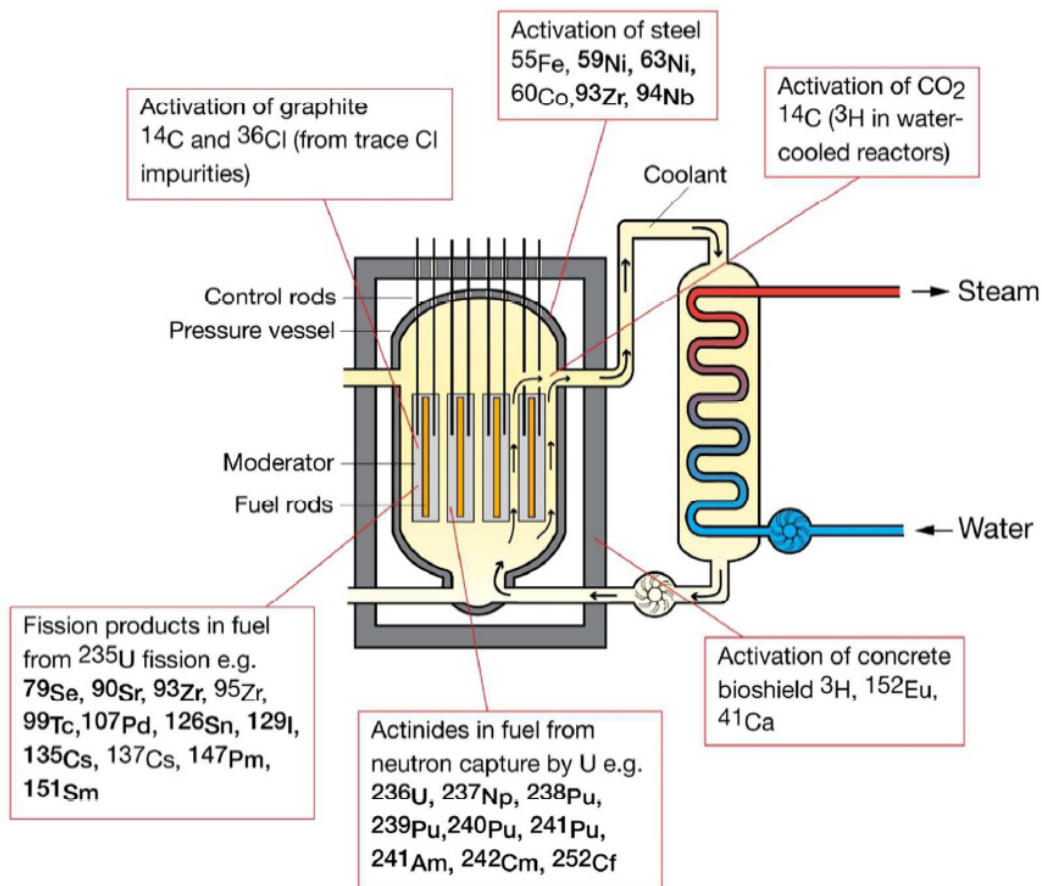


- ✓ **First PWR** to be built in Europe (10MW(e))
- ✓ Located at SCK•CEN
- ✓ Used for 25 years for research purposes (*testing high burn-up and gadolinium type fuels and also testing of the mixed oxide fuels*)
- ✓ Shut down in 1987
- ✓ In 1989 was selected by the EC as one of the 4 **pilot dismantling projects**

Surface repository

- In Belgium, a near surface repository for low and intermediate radioactive waste is planned for construction
- Once the repository is sealed, a monitoring plan will be implemented in order to ensure the safety of the environment and the population
- Sound environmental monitoring of nuclear sites or waste repositories also requires the measurement of the radionuclides at acceptable radioactivity concentrations

“Difficult to measure” radionuclides



- ✓ Produced during activation of different parts of a nuclear reactor
- ✓ Also during the fission of ^{235}U
- ✓ Very important for **classification of the materials** from nuclear installations
- ✓ since they are *mainly pure alpha and beta emitters* are not easy to be quantified

How to characterize the materials resulted during decommissioning?

Combination of various characterization techniques

- ✓ Destructive Analyses (DA) ➡ **'Difficult To Measure'**
- ✓ Non Destructive Assay (NDA) ➡ Easy To Measure
- ✓ Modelling

Scaling factors

- ✓ The quantification of the pure alpha, beta-emitters is time consuming and expensive, so it is not feasible to perform destructive analysis on each waste batch
- ✓ Therefore, scaling factors are used
- ✓ Scaling factors relate the activity of the critical radionuclides to **Cs-137, for fission products**, and to **Co-60 for activation products**.
- ✓ Both Cs-137 and Co-60 are gamma emitters that can be easily measured in waste using non-destructive methods.
- ✓ The scaling factors can be estimated using **calculation codes**
- ✓ This calculation codes needs to be **validated**
- ✓ So, **real analyses** have to be performed on different types of samples

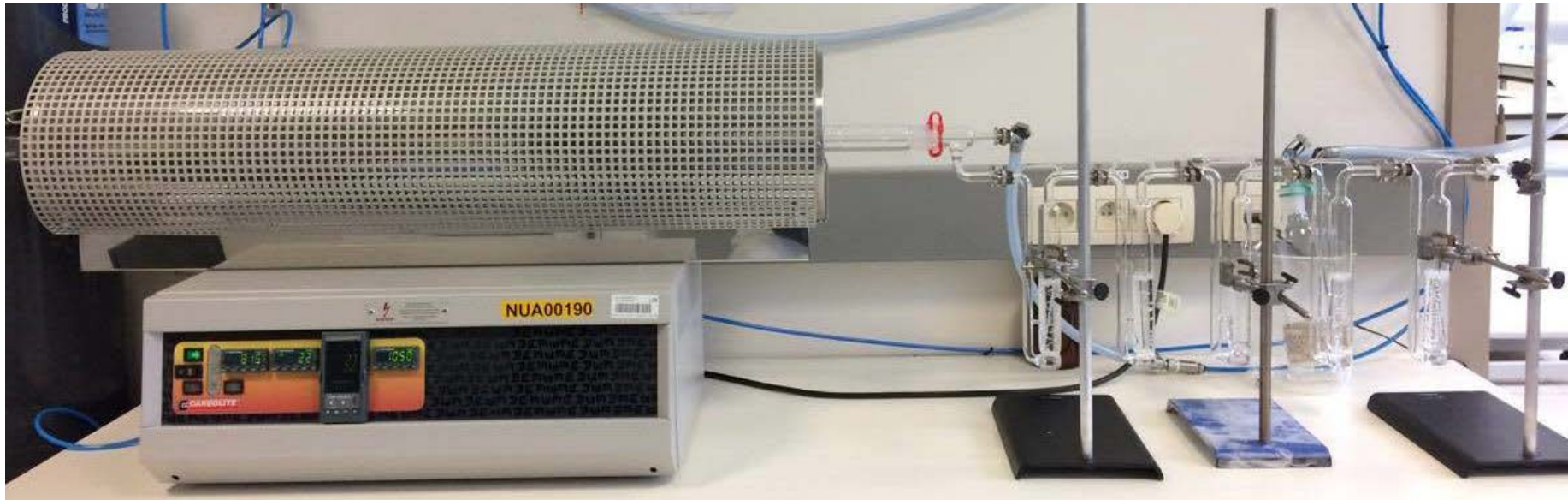
- Focus was:
 - in optimizing sample preparation for difficult matrices, such as: concrete, resins, evaporator concentrates
 - optimizing separation of actinides using different methods
 - optimizing the measurement methods (alpha-spectrometry, mass spectrometry, LSC)

- SCK•CEN + EC project 1999 + **actinides in difficult matrices**
- WP1 Dissolution methods
 - *Microwave dissolution*
 - *Leaching*
 - *Fusion*
- WP2 Separation methods
 - *Ion exchange*
 - *Extraction chromatography*
- WP3 Measurement methods (source preparation)
- ...

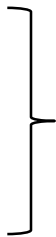
Current challenges

- RCA (high active samples)+ LRM (low active samples) collaboration for determination of other radionuclides
- actinides + ^{14}C , ^{36}Cl , ^{94}Nb , ^{99}Tc , ^{129}I , ^{63}Ni , ^{55}Fe
- Different types of matrices: resins, evaporator concentrates, cemented waste, environmental samples
- Mass spectrometry + LSC

Determination of ^{14}C , ^{36}Cl , ^{129}I



- ✓ ^{14}C
- ✓ ^{36}Cl
- ✓ ^{129}I



- ✓ Graphite powder
- ✓ filter paper



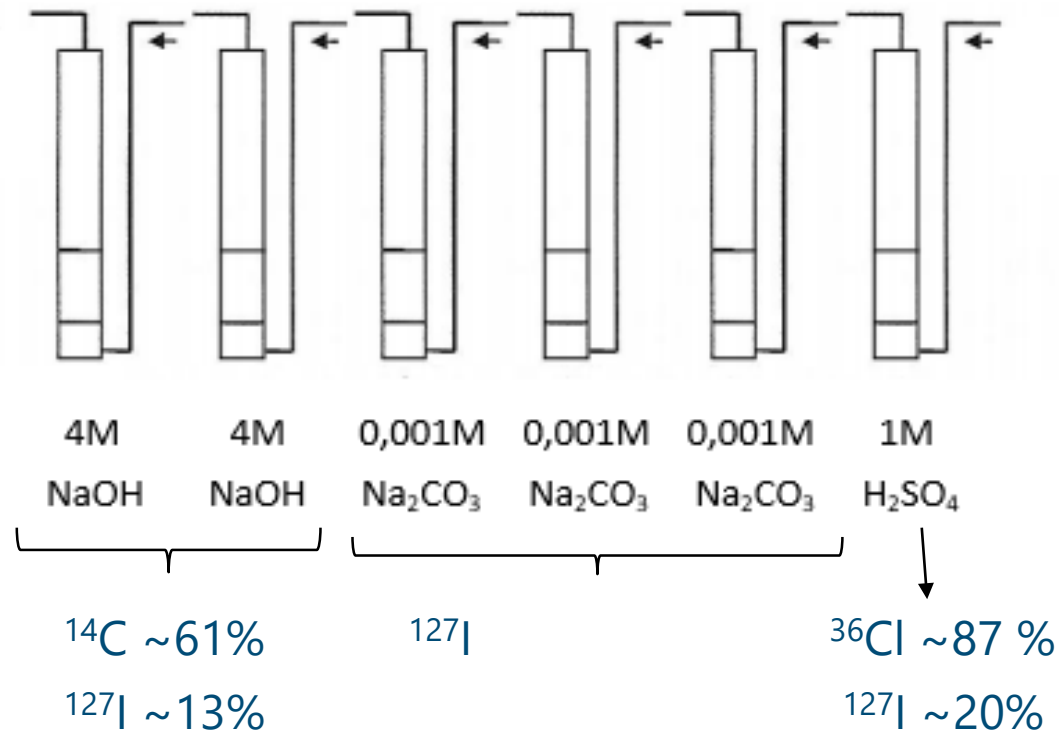
Challenges:

- ✓ Separation of the 3 radionuclides in one combustion run
- ✓ Good repeatability
- ✓ High absorption yields

Determination of ^{14}C , ^{36}Cl , ^{129}I

Challenges:

- ✓ Separation of the 3 radionuclides in one combustion run - ok
- ✓ Good repeatability – relatively ok
- ✓ **But Iodine was present in other fractions as well**



Proposed investigations/solutions?

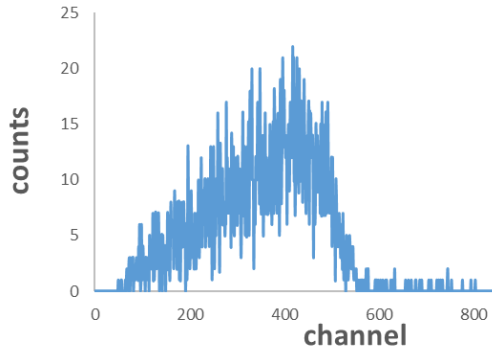
- ✓ $^{36}\text{Cl} + ^{129}\text{I}$ can be measured together by LSC – separation using Cl-resin – results not reliable
- ✓ $^{14}\text{C} + ^{129}\text{I}$ is a problem – new separation by acidolize and then ^{129}I measured by ICP-MS
- ✓ keep on looking for a simpler and faster solution

⁹⁹Tc quantification in environmental samples

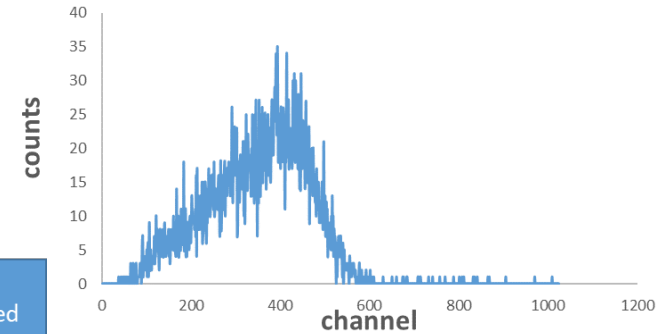
- ⁹⁹Tc quantification in environmental samples
- $E_{\beta_{max}} = 293.8 \text{ keV}$
- Liquid scintillation counting as main measuring technique
- TEVA resin is used for its radiochemical separation from solid samples

^{99}Tc determination using Tc RAD discs

^{99}Tc : sand + spike (dried)



^{99}Tc : sand + spike (diluted)



Solid sample
(microwave digestion using
 HNO_3 , H_2O_2 , HF)

Sample evaporated to dryness;
residue dissolved in conc. HCl
and diluted to 250 mL

or

conc. HCl added to the
dissolved sample and diluted
to 250 mL

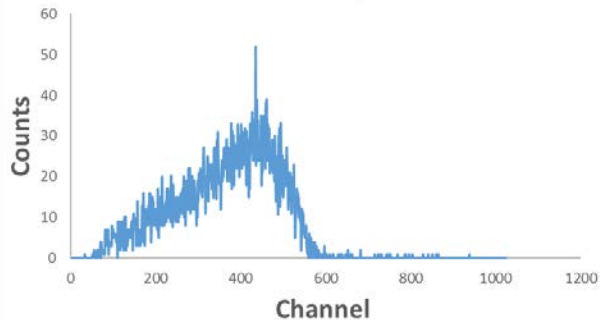
Water sample
(0.012 M HCl)

Filter through 3M
Empore™ Technetium
RAD Disk

Disk + 20 mL
Optiphase HiSafe III

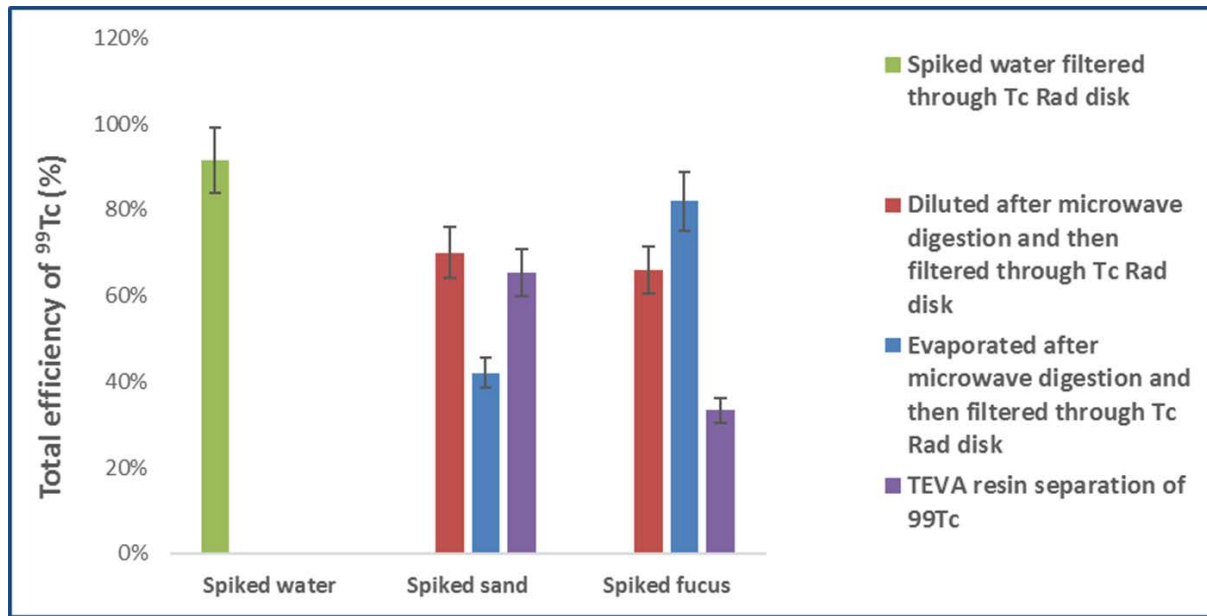
Quantulus
(full spectrum)

^{99}Tc : water + spike



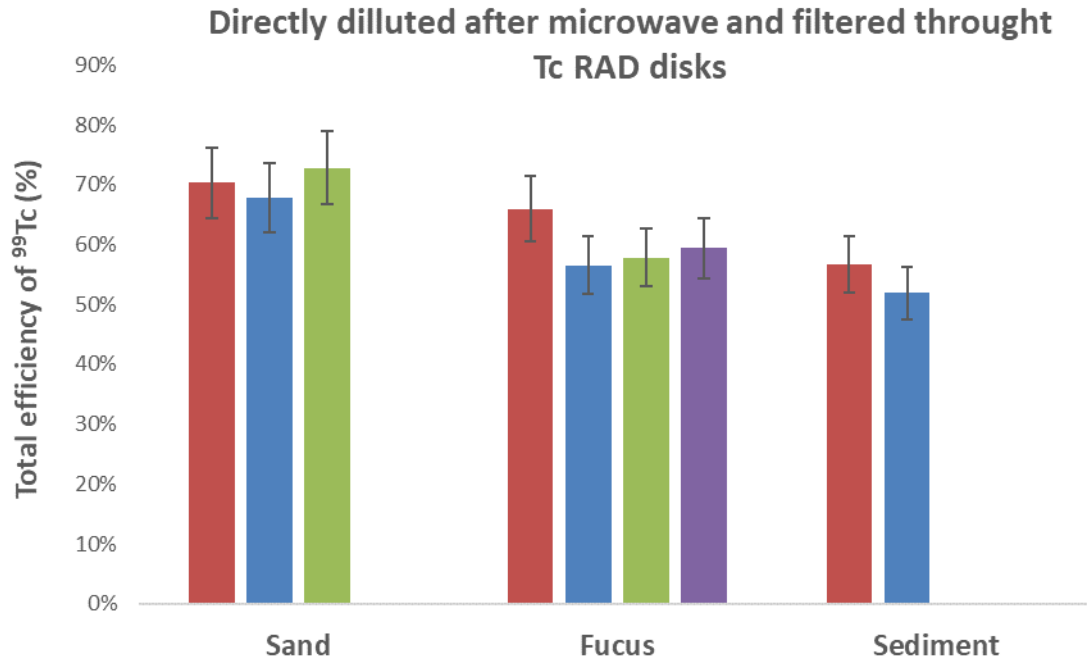
^{99}Tc determination using Tc RAD discs

- testing the **Technetium RAD disks** for solid samples



- ✓ **Total efficiency** of ^{99}Tc was influenced by the **sample matrix**, lower values were obtained using spiked sand sample comparing with water sample
- ✓ **Dilution of the sample after digestion** with microwave, then filtration through Tc Rad disk **seems to be promising** and it makes the procedure very simple and much faster comparing with the TEVA resin separation

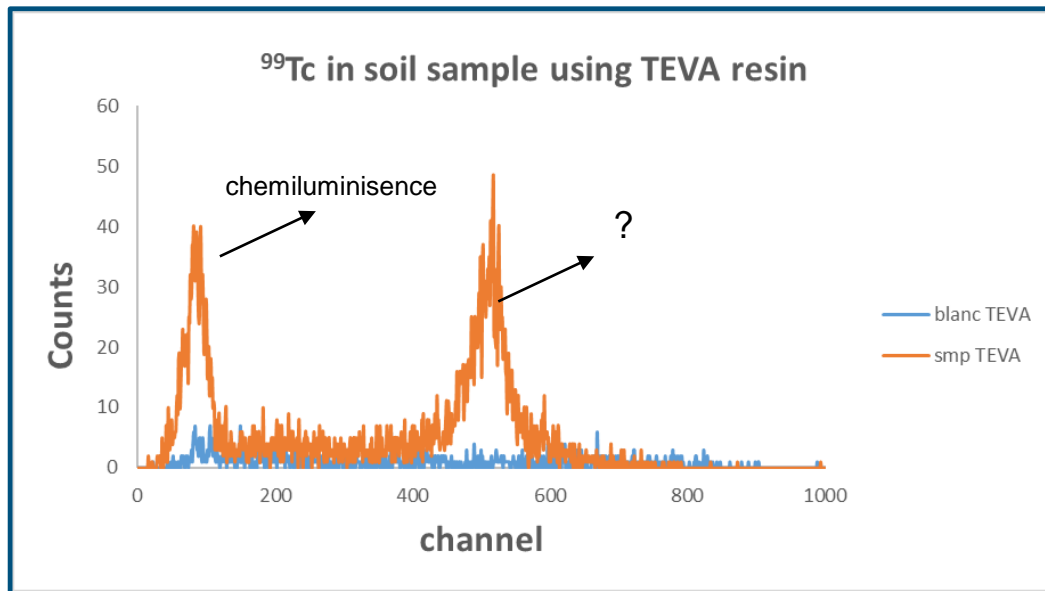
^{99}Tc determination using Tc RAD discs



- ✓ **Dilution of the sample after digestion** with microwave, then filtration through Tc Rad disk **seems to be promising** and it makes the procedure very simple and much faster comparing with the TEVA resin separation
- ✓ The total efficiency is between **60 – 70 %**

^{99}Tc determination using TEVA resin

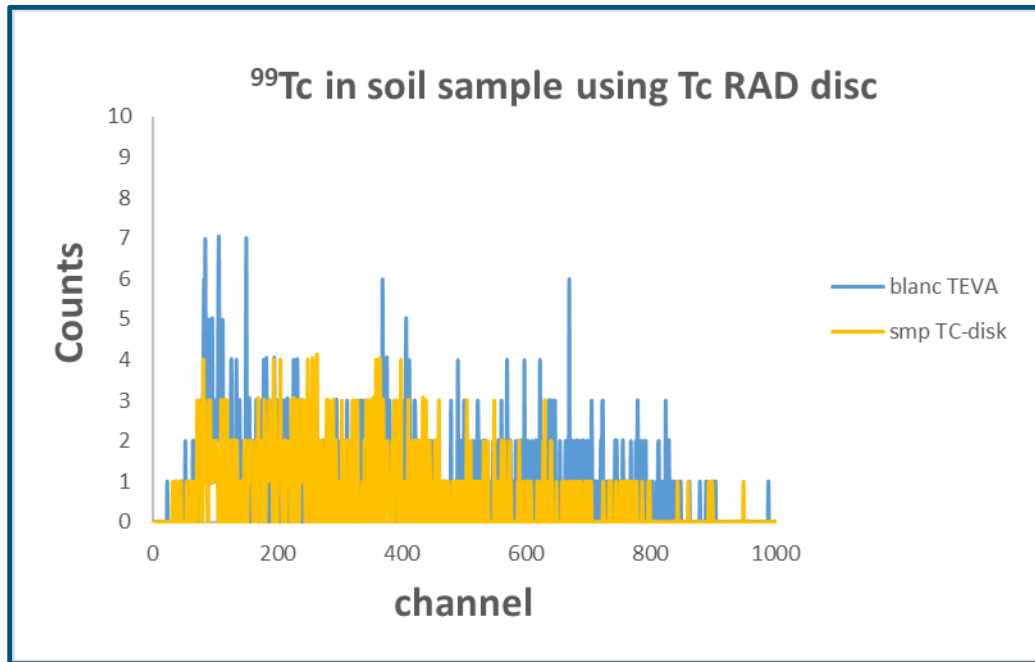
Sediment sample: ^{99}Tc separated using TEVA resin and Quantulus1220™ for its quantification



- ✓ an extra peak in the spectrum
- ✓ overestimation of the massic activity
- ✓ separation not ok?
- ✓ cross-contamination?

^{99}Tc determination using Tc RAD discs

Sediment sample: ^{99}Tc separated using TEVA resin and Quantulus1220™ for its quantification



- ✓ **No** extra peak in the spectrum
- ✓ Good separation

- ✓ **3M decided to stop with the production of the discs**
- **Perspectives/options to be investigated in the future:**
 - ✓ TEVA discs from Eichrom
 - ✓ Plastic scintillators

- **PhD topic:** "Development of analysis methods for '*difficult to measure*' radionuclides in environmental samples around nuclear sites and in materials produced during decommissioning activities"
- **^{79}Se , ^{36}Cl , ^{129}I , ^{151}Sm , ^{147}Pm , ^{41}Ca**
- Matrices: environmental and/or nuclear materials produced during decommissioning activities (using fusion)
- LSC (Liquid Scintillation Counting)+ICP-QQQMS

^{79}Se ($E_{\beta\text{max}} = 150.9 \text{ keV}$) - After a very laborious separation procedure to remove the spectrometric interferences

- LSC is the usual quantitative measurement technique employed
 - The current detection limits achievable by mass spectrometry are much higher due to isobaric interference
 - The ICP-QQQ-MS technique opens up the potential for using reaction chemistries
-
- **^{36}Cl** ($E_{\beta\text{max}} = 709.6 \text{ keV}$)
 - often measured by LSC
 - Cl-resin has been developed by Triskem
 - the use of the Pyrolyzer and Cl-resin

- **^{151}Sm** ($E_{\beta\text{max}} = 76.4 \text{ keV}$) and **^{147}Pm** ($E_{\beta\text{max}} = 224 \text{ keV}$)
 - The use of the extraction chromatography (Ln-resin) has been reported in the literature
 - Liquid scintillation counting will be the main quantitative technique
- **^{41}Ca** (421.6 keV)
 - Very complicated and lengthy separation procedures are employed for removal of the interferences.
 - Since ^{41}Ca decays by electron capture, its quantification is achieved by LSC or X-ray spectrometry.

- a lot has been done in the past
- with new safety studies new nuclides of concern are considered and need to be measured
- so techniques need to be developed (and accredited when it comes to making official analyses)

Thank you for your attention!

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